

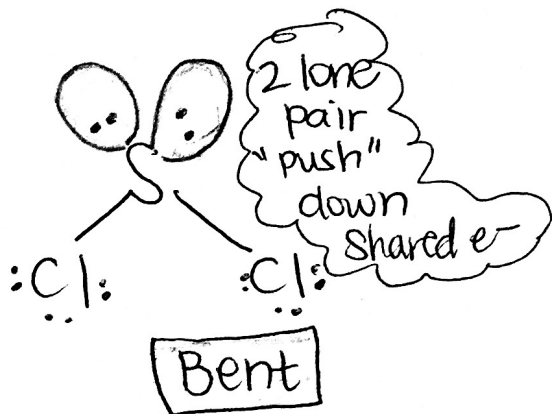
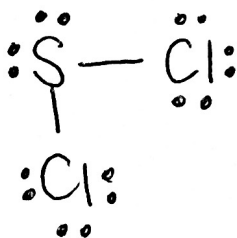
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a. SCl_2

S = 6 valence e^-

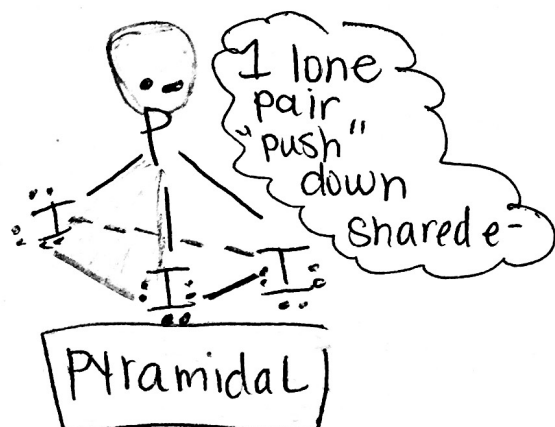
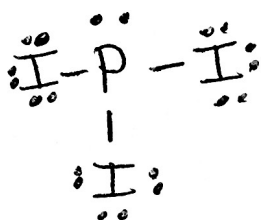
Cl = 7 valence e^-



b. PI_3

P = 5 valence e^-

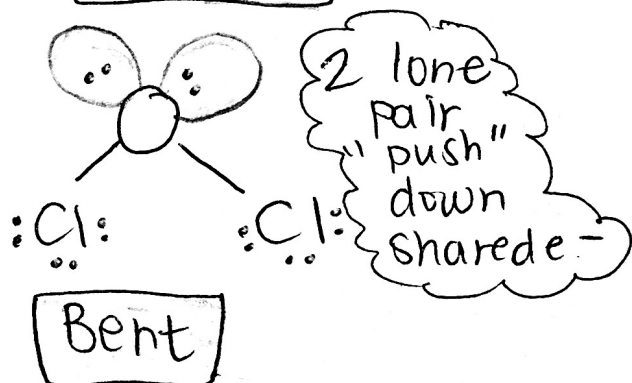
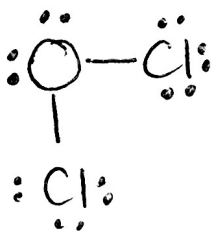
I = 7 valence e^-



c. Cl_2O

O = 6 valence e^-

Cl = 7 valence e^-

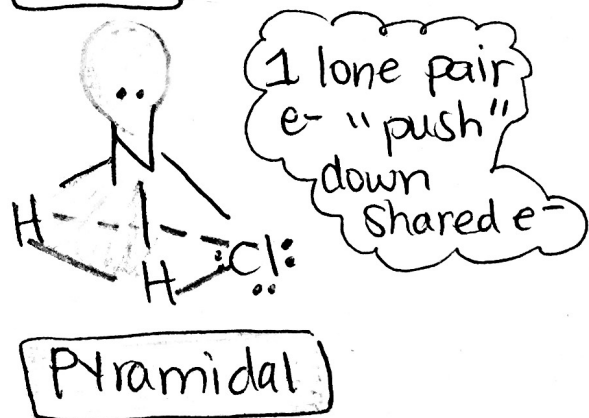
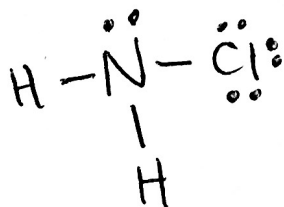


d. NH_2Cl

N = 5 valence e^-

H = 1 valence e^-

Cl = 7 valence e^-

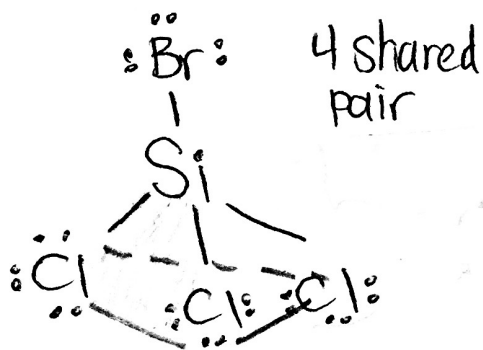
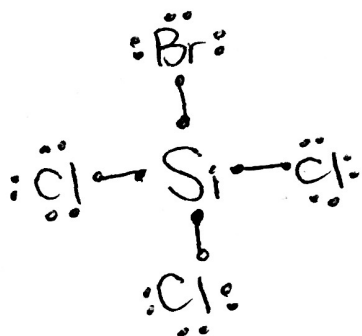




Si = 4 valence e^-

Cl = 7 valence e^-

Br = 7 valence e^-



Tetra hedral