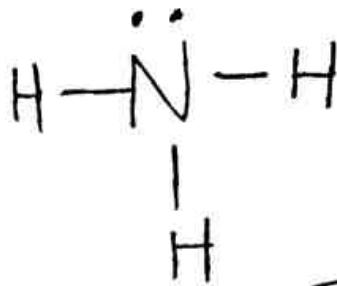
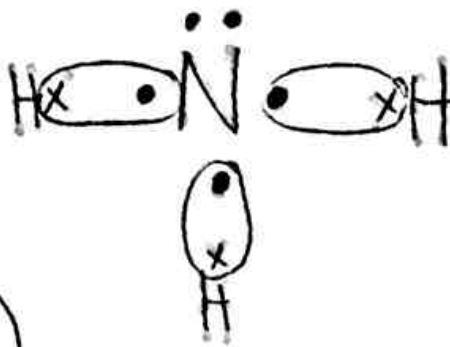
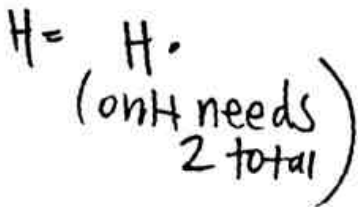
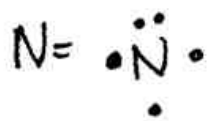
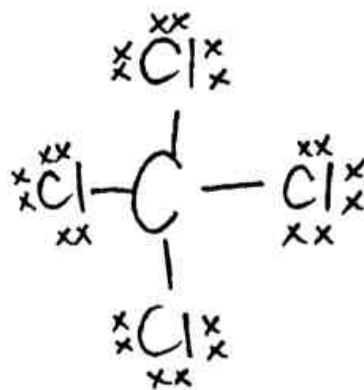
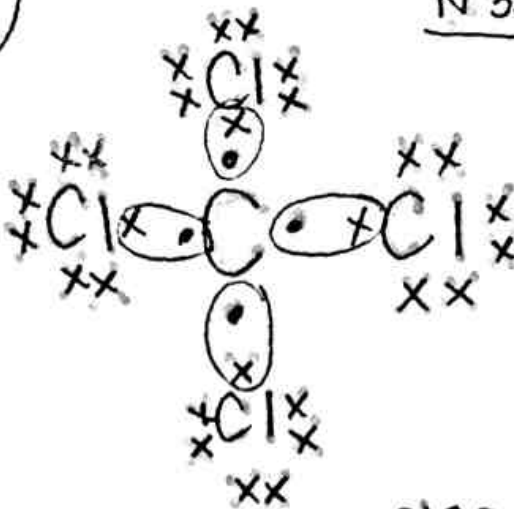
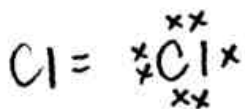
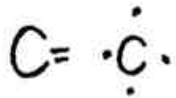


1. NH_3



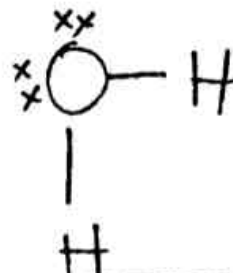
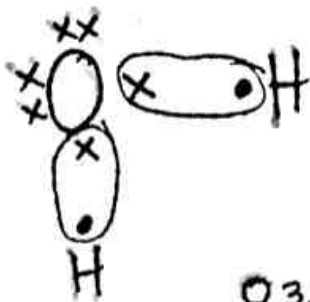
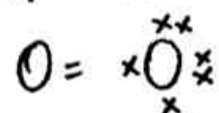
$\text{N } 3.0 - 2.1 \text{ H} = 0.9 \text{ polar Bond}$

2. CCl_4



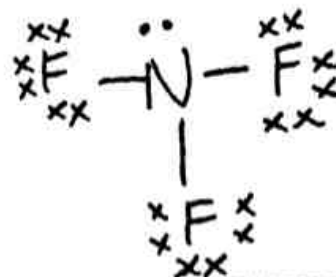
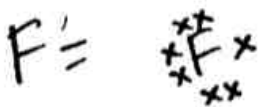
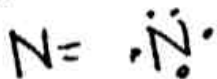
$\text{Cl } 3.0 - 2.5 \text{ C} = 0.5 \text{ polar Bond}$

3. H_2O



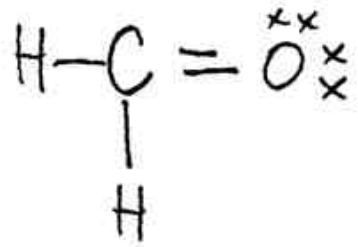
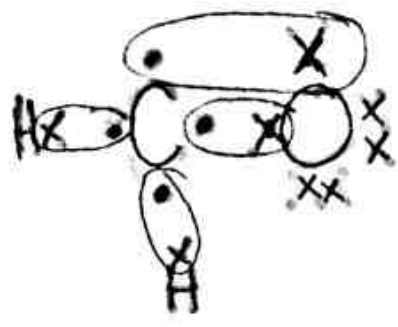
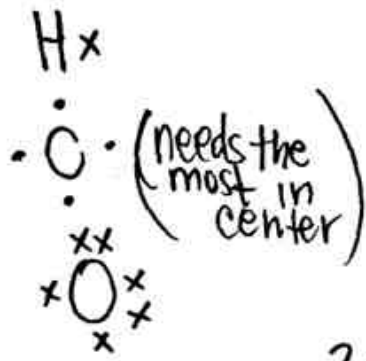
$\text{O } 3.5 - 2.1 \text{ H} = 1.4 \text{ polar Bond}$

1. NF_3



$\text{F } 4.0 - 3.0 \text{ N} = 1.0 \text{ polar bond}$

5. H_2CO

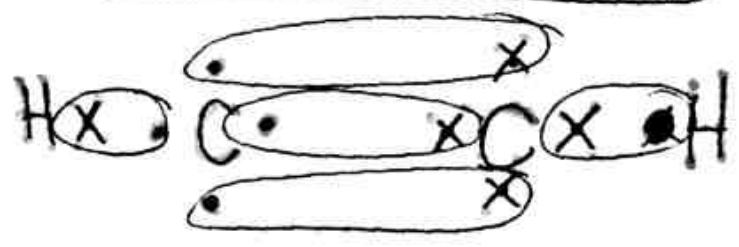
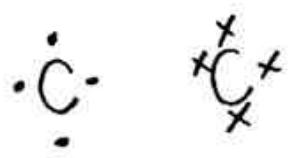


2 different bonds:

$$C 2.5 - 2.1 H = 0.4 \text{ polar bond}$$

$$O 3.5 - 2.5 C = 1.0 \text{ polar bond}$$

6. C_2H_2



$$C 2.5 - 2.1 H = 0.4 \text{ polar bond}$$